

## Natural Products

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## Total Synthesis of IKD-8344\*\*

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IKD-8344 (**1**, Scheme 1) is a novel 28-membered ring macrodiolide antibiotic from an unidentified alkalophilic Actinomycete, strain no. 8344,<sup>[1]</sup> which exhibits potent anthelmintic activity against *Trichinella spiralis* and strong cytotoxicity against L5178Y mouse leukemia cells with an IC<sub>50</sub> value of 0.54 ng mL<sup>-1</sup>.<sup>[1]</sup> More recently, it was also isolated from *Streptomyces* sp. A6792 and selective antifungal activities against the mycelial form of *Candida albicans* were noted.<sup>[2]</sup>

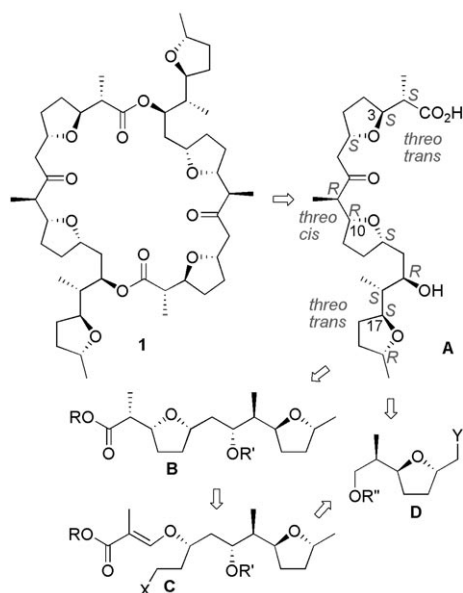
The unique bioactivities and stereochemical complexities of the molecule make it an attractive synthetic target. The monomeric *seco* acid **A** (Scheme 1) contains three methyl branches and three oxolane units: a *threo* (C2–C3)–*trans* (C3–C6) array is connected to a *threo* (C9–C10)–*cis* (C10–C13) arrangement, which is flanked by another *threo* (C16–C17)–*trans* (C17–C20) unit. Efficient and stereoselective construction and assemblage of these structural units is not trivial, and

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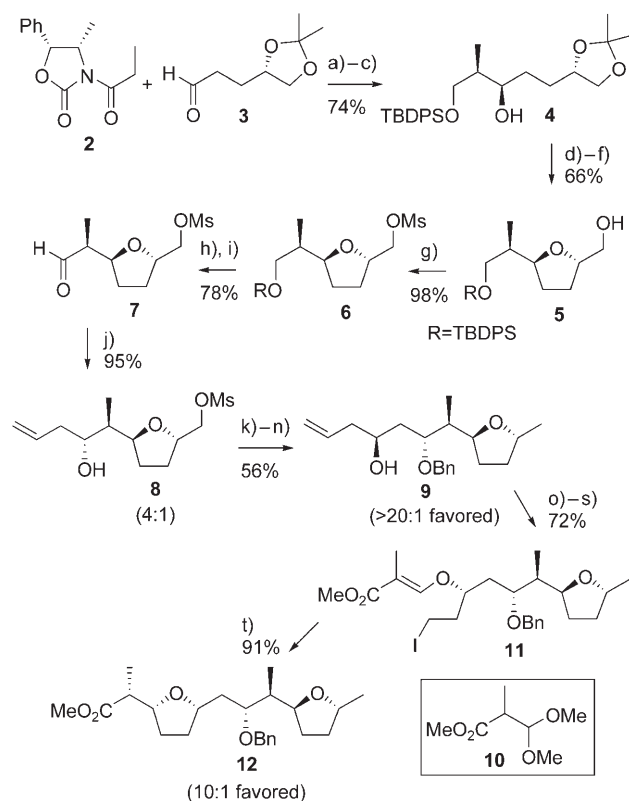


**Scheme 1.** Retrosynthetic analysis.

the difficulty therein is manifested by the absence of reports on the total synthesis of IKD-8344 (**1**) in the literature.<sup>[3]</sup> In the retrosynthetic analysis, synthesis of **1** would be accomplished through dimerization of the monomeric *seco* acid **A**, which may be prepared by coupling of fragments **B** and **D**. Fragment **B** may be obtained by radical cyclization of the  $\beta$ -alkoxymethacrylate derivative **C**, which should be accessible from fragment **D**. A reliable synthesis of fragment **D** is prerequisite for a successful total synthesis of this intriguing molecule, which is described in this communication.

Synthesis of the **B** fragment commenced with the aldol reaction of aldehyde **3**<sup>[4]</sup> with the boron enolate of the chiral imide **2** (Scheme 2).<sup>[5]</sup> Reduction of the product aldol imide and silyl protection of the primary hydroxy group afforded alcohol **4**. The *threo*–*trans* oxolane derivative **5** was obtained by mesylation of **4**, acetonide deprotection, and treatment with base. The corresponding mesylate **6** was converted into aldehyde **7** by silyl deprotection and PCC oxidation. Reaction of **7** with allyltrimethylsilane in the presence of titanium chloride led to a 4:1 mixture,<sup>[6]</sup> favoring **8**, of the homoallylic alcohols. At this point, the mesyloxy group was removed through  $\text{LiAlH}_4$  reduction, and the aldehyde obtained by ozonolysis after benzyl protection was converted stereoselectively into the homoallylic alcohol **9** by reaction with allyltributylstannane in the presence of magnesium bromide etherate.<sup>[7]</sup> Iodide **11** was obtained from **9** by ozonolysis, sodium borohydride reduction, tosylation of the primary hydroxy group, reaction with the dimethyl acetal **10**,<sup>[8]</sup> and iodide substitution. Radical cyclization<sup>[9]</sup> of  $\beta$ -alkoxymethacrylate **11** proceeded smoothly to produce selectively (10:1) the *threo*–*cis* oxolane product **12** in good yield (Scheme 2).<sup>[10]</sup>

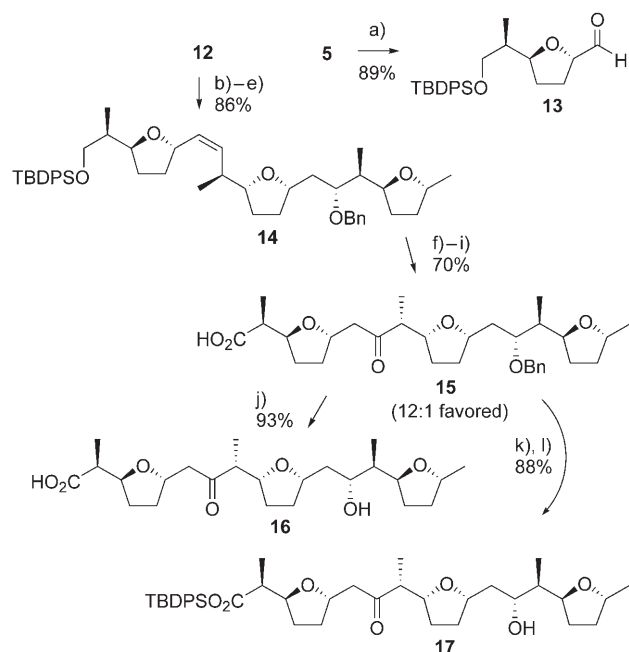
Coupling of fragments **B** and **D** was problematic. For example, the dithiane derivative prepared from ester **12** was not amenable to the reaction with mesylate **6**. After considerable exploration, it was found that efficient coupling was possible by employing a Wittig reaction. Aldehyde **13** was



**Scheme 2.** Preparation of the **B** fragment. a) **2**,  $\text{Bu}_2\text{BOTf}$ ,  $\text{NEt}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-40 \rightarrow 0^\circ\text{C}$ ; **3**,  $-78^\circ\text{C}$ ; b)  $\text{LiBH}_4$ , diethyl ether,  $0^\circ\text{C} \rightarrow \text{RT}$ ; c)  $\text{TBDPSCl}$ , imidazole,  $\text{CH}_2\text{Cl}_2$ ; d)  $\text{MsCl}$ ,  $\text{NEt}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C} \rightarrow \text{RT}$ ; e)  $\text{CSA}$ ,  $\text{MeOH}$ ; f)  $\text{NaHMDS}$ ,  $\text{THF}$ ,  $0^\circ\text{C} \rightarrow \text{RT}$ ; g)  $\text{MsCl}$ ,  $\text{NEt}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C} \rightarrow \text{RT}$ ; h)  $\text{TBAF}$ ,  $\text{THF}$ ; i)  $\text{PCC}$ , 4-Å  $\text{MS}$ ,  $\text{CH}_2\text{Cl}_2$ ; j)  $\text{CH}_2\text{CHCH}_2\text{SiMe}_3$ ,  $\text{TiCl}_4$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-78^\circ\text{C}$ ; k)  $\text{LiAlH}_4$ , diethyl ether,  $0^\circ\text{C} \rightarrow \text{RT}$ ; l)  $\text{NaHMDS}$ ,  $\text{BnBr}$ ,  $\text{THF}/\text{DMF}$  (5:1),  $0^\circ\text{C} \rightarrow \text{RT}$ ; m)  $\text{O}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-78^\circ\text{C}$ ;  $\text{Ph}_3\text{P}$ ; n)  $\text{CH}_2\text{CHCH}_2\text{SnBu}_3$ ,  $\text{MgBr}_2 \cdot \text{Et}_2\text{O}$ ,  $\text{CH}_2\text{Cl}_2$ ; o)  $\text{O}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-78^\circ\text{C}$ ;  $\text{Ph}_3\text{P}$ ; p)  $\text{NaBH}_4$ ,  $\text{MeOH}$ ,  $0^\circ\text{C} \rightarrow \text{RT}$ ; q)  $\text{TsCl}$ ,  $\text{NEt}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C}$ ; r) **10**,  $\text{PPTS}$ , benzene, reflux; s)  $\text{NaI}$ , acetone, reflux; t)  $\text{Bu}_3\text{SnH}$ ,  $\text{Et}_3\text{B}$ , toluene, air,  $-78^\circ\text{C}$ .  $\text{Tf}$  = trifluoromethanesulfonyl,  $\text{TBDPS}$  = *tert*-butyldiphenylsilyl,  $\text{Ms}$  = methanesulfonyl,  $\text{CSA}$  = ( $\pm$ )-camphorsulfonic acid,  $\text{HMDS}$  = 1,1,1,3,3,3-hexamethyldisilazane,  $\text{TBAF}$  = tetrabutylammonium fluoride,  $\text{PCC}$  = pyridinium chlorochromate,  $\text{Bn}$  = benzyl,  $\text{Ts}$  = toluene-4-sulfonyl,  $\text{PPTS}$  = pyridinium *p*-toluenesulfonate.

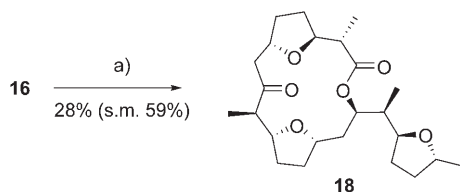
prepared from the primary alcohol **5** (Scheme 3). Ester **12** was converted into the corresponding phosphonium salt by lithium borohydride reduction, iodide substitution, and reaction with triphenylphosphine, and the ylide obtained from the phosphonium salt reacted with aldehyde **13** to produce olefin **14** in good yield. A mixture of epoxides was obtained from **14** by oxidation with *m*CPBA. Reaction of the mixture with excess lithium aluminum hydride, Dess–Martin oxidation of the diol produced, and chlorite oxidation of the aldehyde functionality provided regioselectively (12:1) the correct keto carboxylic acid **15**. Use of diisobutylaluminum hydride ( $\text{DIBALH}$ ) as the reducing agent eventually produced the alternative ketone as the major product. The monomeric *seco* acid **16** was prepared by hydrogenolysis. The secondary alcohol **17** was obtained from **15** by  $\text{TBDPS}$  protection and hydrogenolysis (Scheme 3).

Direct macrodiolide formation of the monomeric *seco* acid **16** did not proceed. A relatively low yield of the



**Scheme 3.** Preparation of the **A** fragment. a)  $\text{SO}_3$ :pyridine,  $\text{NEt}_3$ ,  $\text{DMSO}/\text{CH}_2\text{Cl}_2$  (1:1); b)  $\text{LiBH}_4$ , diethyl ether,  $0^\circ\text{C} \rightarrow \text{RT}$ ; c)  $\text{I}_2$ ,  $\text{Ph}_3\text{P}$ , imidazole, THF,  $0^\circ\text{C} \rightarrow \text{RT}$ ; d)  $\text{Ph}_3\text{P}$ , MeCN, reflux; e)  $\text{BuLi}$ , THF,  $-78^\circ\text{C}$ ; **13**; f)  $m\text{CPBA}$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C}$ ; g)  $\text{LiAlH}_4$ , diethyl ether,  $0^\circ\text{C}$ ; h) Dess–Martin periodinane (3.0 equiv),  $\text{CH}_2\text{Cl}_2$ ; i)  $\text{NaClO}_2$  (1.3 equiv),  $\text{NaH}_2\text{PO}_4$  (1.3 equiv),  $t\text{BuOH}/2\text{-methyl-2-butene}/\text{H}_2\text{O}$  (10:5:1); j)  $\text{H}_2$ ,  $\text{Pd}/\text{C}$ , MeOH; k)  $\text{TBDPSCl}$  (4.0 equiv), imidazole (7.0 equiv), DMAP (0.5 equiv),  $\text{CH}_2\text{Cl}_2$ ; l)  $\text{H}_2$ ,  $\text{Pd}(\text{OH})_2/\text{C}$ , EtOAc.  $m\text{CPBA}$  = 3-chloroperoxybenzoic acid, DMAP = 4-dimethylaminopyridine.

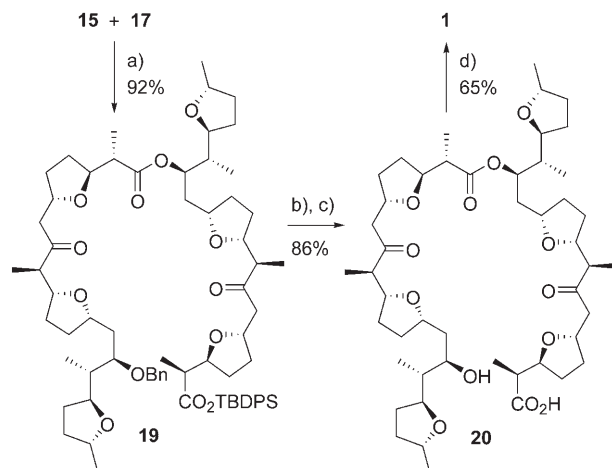
macrolide product **18** was obtained under a variety of conditions (Scheme 4).<sup>[11]</sup> The coupling reaction of **15** and **17**, however, proceeded efficiently under standard Yamaguchi



**Scheme 4.** Cyclization of the *seco* acid. a) CDIC (2.5 equiv),  $\text{Cs}_2\text{CO}_3$  (1.0 equiv),  $\text{CH}_2\text{Cl}_2$  (0.02 M),  $0^\circ\text{C}$ ; DMAP (20 equiv). CDIC = 2-chloro-1,3-dimethylimidazolium chloride, s.m. = recovered starting material.

conditions to afford ester **19** (Scheme 5). The dimeric *seco* acid **20** was obtained efficiently through TBDPS deprotection and hydrogenolysis. IKD-8344 (**1**)<sup>[12]</sup> was finally obtained upon lactonization of **20** under modified Yamaguchi conditions (Scheme 5).<sup>[13]</sup>

The present synthesis is another example of the application of  $\beta$ -alkoxymethacrylate radical cyclization reactions<sup>[9]</sup> for the stereoselective construction of complex oxacyclic natural products.



**Scheme 5.** Synthesis of IKD-8344. a) **15** (1.0 equiv), 2,4,6- $\text{Cl}_3\text{PhCOCl}$  (1.7 equiv),  $\text{NEt}_3$  (3.5 equiv), THF (0.025 M); **17** (1.3 equiv), DMAP (3.0 equiv), benzene (0.02 M); b) conc.  $\text{HCl}$ , MeOH; c)  $\text{H}_2$ ,  $\text{Pd}(\text{OH})_2/\text{C}$ , EtOAc; d) 2,4,6- $\text{Cl}_3\text{PhCOCl}$  (40 equiv);  $\text{NEt}_3$  (60 equiv), DMAP (20 equiv), toluene (0.15  $\mu\text{M}$ ), reflux.

## Experimental Section

Ester **12**:  $n\text{Bu}_3\text{SnH}$  (0.018 mL, 0.07 mmol) and  $\text{Et}_3\text{B}$  (1.0 M in hexane, 0.088 mL) were added to a solution of iodide **11** (31 mg, 0.058 mmol) in toluene (6 mL) at  $-78^\circ\text{C}$ . The resulting solution was stirred for 1 h under air at  $-78^\circ\text{C}$ . Concentration and purification of the residue by flash column chromatography (hexanes/EtOAc, 4:1) furnished ester **12** (21.5 mg, 91%);  $R_f$  = 0.31 (hexanes/EtOAc, 4:1);  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 7.33–7.24 (m, 5H), 4.53 and 4.47 (AB q, 2H,  $J_{\text{AB}}$  = 11.3 Hz), 4.08–3.99 (m, 3H), 3.86–3.78 (m, 2H), 3.69 (s, 3H), 2.60–2.55 (m, 1H), 2.05–1.90 (m, 6H), 1.70–1.39 (m, 5H), 1.19 (d, 3H,  $J$  = 6.0 Hz), 1.11 (d, 3H,  $J$  = 7.0 Hz), 0.82 ppm (d, 3H,  $J$  = 6.9 Hz);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 175.5, 139.2, 128.2, 127.7, 127.2, 80.0, 79.6, 77.9, 77.2, 74.6, 71.6, 51.6, 45.2, 40.6, 36.5, 33.9, 31.6, 30.7, 28.4, 21.4, 13.1, 10.0 ppm; IR (neat):  $\nu_{\text{max}}$  = 3029, 2967, 2877, 1739, 1455, 1376, 1197, 1068, 736, 698  $\text{cm}^{-1}$ ; MS (CI):  $m/z$  (relative intensity): 405 [ $M^+ + 1$ ] (81), 297 (100), 295 (28), 279 (11), 213 (63), 183 (31), 157 (42), 91 (23); HRMS (CI): calcd for  $\text{C}_{24}\text{H}_{37}\text{O}_5$  [ $M^+ + 1$ ]: 405.2641; found: 405.2641;  $[\alpha]_D^{25}$  = +14.0 ( $c$  = 1.16,  $\text{CHCl}_3$ ).

IKD-8344 (**1**): The dimeric *seco* acid **20** (15 mg, 0.017 mmol) was dissolved in toluene (113 mL).  $\text{NEt}_3$  (0.15 mL, 1.04 mmol), 2,4,6-trichlorobenzoyl chloride (0.11 mL, 0.70 mmol), and DMAP (42 mg, 0.35 mmol) were added at room temperature, and then the solution was heated under reflux for 4 h. The reaction was quenched by addition of saturated  $\text{NH}_4\text{Cl}$  solution (100 mL) and the reaction mixture was extracted with  $\text{CH}_2\text{Cl}_2$  (150 mL  $\times$  3). The organic extracts were dried over  $\text{MgSO}_4$ , filtered, and concentrated. Purification of the residue by flash column chromatography (hexanes/EtOAc, 2:1) gave IKD-8344 (**1**, 9.5 mg, 65%);  $R_f$  = 0.28 (hexanes/EtOAc, 2:1);  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 5.32–5.28 (m, 2H), 4.38–4.31 (m, 2H), 4.15–4.01 (m, 4H), 3.96–3.87 (m, 4H), 3.82–3.76 (m, 2H), 2.95–2.83 (m, 4H), 2.46–2.26 (m, 6H), 2.19–2.13 (m, 2H), 2.03–1.90 (m, 9H), 1.81–1.73 (m, 3H), 1.68–1.59 (m, 5H), 1.51–1.38 (m, 9H), 1.18 (d, 6H,  $J$  = 6.0 Hz), 1.09 (d, 6H,  $J$  = 7.4 Hz), 1.00 (d, 6H,  $J$  = 7.3 Hz), 0.83 ppm (d, 6H,  $J$  = 7.5 Hz);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 212.0, 175.1, 80.5, 80.0, 79.5, 75.2, 74.9, 74.8, 72.2, 53.4, 45.6, 45.6, 41.1, 36.2, 33.9, 32.1, 30.9, 30.7, 29.6, 29.4, 21.2, 14.4, 13.7, 10.9 ppm; MS (FAB):  $m/z$  (relative intensity): 867 [ $M^+ + \text{Na}$ ] (58), 883 (45), 845 (14), 423 (12), 307 (22), 289 (11), 209 (43), 154 (83), 85 (100); HRMS (FAB): calcd for  $\text{C}_{48}\text{H}_{76}\text{O}_{12}\text{Na}$  [ $M^+ + \text{Na}$ ]: 867.5235; found: 867.5239;  $[\alpha]_D^{15}$  = +39.7 ( $c$  = 0.25,  $\text{CHCl}_3$ ).

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- [1] a) Y. Minami, K. Yoshida, R. Azuma, M. Nishii, J. Inagaki, F. Nohara, *Tetrahedron Lett.* **1992**, 33, 7373–7376; b) T. Ishida, Y. In, M. Nishii, Y. Minami, *Chem. Lett.* **1994**, 1321–1322.
- [2] E. I. Hwang, B. S. Yun, W. H. Yeo, S. H. Lee, J. S. Moon, Y. K. Kim, S. J. Lim, S. U. Kim, *J. Microbiol. Biotechnol.* **2005**, 15, 909–912.
- [3] For a report on the synthesis of the fragments, see: W. Jiang, D. A. Lantrip, P. L. Fuchs, *Org. Lett.* **2000**, 2, 2181–2184.
- [4] For an example of the use of this known aldehyde, see: A. B. Smith III, D.-S. Kim, *J. Org. Chem.* **2006**, 71, 2547–2557.
- [5] For an example of asymmetric aldol reactions, see: D. A. Evans, S. W. Kaldor, T. K. Jones, J. Clardy, T. J. Stout, *J. Am. Chem. Soc.* **1990**, 112, 7001–7031.
- [6] Formation of the desired *anti-anti* product was problematic. For example, Roush allylation by using isopropyl (*S,S*)-tartrate yielded a mixture of the products favoring (1:2) the wrong isomer. For detailed discussions, see: D. A. Evans, B. D. Allison, M. G. Yang, C. E. Masse, *J. Am. Chem. Soc.* **2001**, 123, 10840–10852.
- [7] For the use of magnesium bromide etherate in an allylstannane reaction, see: E. Lee, E. J. Jeong, E. J. Kang, L. T. Sung, S. K. Hong, *J. Am. Chem. Soc.* **2001**, 123, 10131–10132.
- [8] R. D. Walkup, N. U. Obeyesekere, *Synthesis* **1987**, 607–611.
- [9] For recent examples of  $\beta$ -alkoxyacrylate and  $\beta$ -alkoxymethacrylate radical cyclization, see: a) E. J. Kang, E. J. Cho, M. K. Ji, Y. E. Lee, D. M. Shin, S. Y. Choi, Y. K. Chung, J.-S. Kim, H.-J. Kim, S.-G. Lee, M. S. Lah, E. Lee, *J. Org. Chem.* **2005**, 70, 6321–6329; b) H. Y. Song, J. M. Joo, J. W. Kang, D.-S. Kim, C.-K. Jung, H. S. Kwak, J. H. Park, E. Lee, C. Y. Hong, S. Jeong, K. Jeon, J. H. Park, *J. Org. Chem.* **2003**, 68, 8080–8087; c) E. J. Jeong, E. J. Kang, L. T. Sung, S. K. Hong, E. Lee, *J. Am. Chem. Soc.* **2002**, 124, 14655–14662; d) E. Lee, S. J. Choi, H. Kim, H. O. Han, Y. K. Kim, S. J. Min, S. H. Son, S. M. Lim, W. S. Jang, *Angew. Chem.* **2002**, 114, 184–186; *Angew. Chem. Int. Ed.* **2002**, 41, 176–178.
- [10] E. Lee, S. J. Choi, *Org. Lett.* **1999**, 1, 1127–1128.
- [11] See, for example: A. Fürstner, J. Mlynarski, M. Albert, *J. Am. Chem. Soc.* **2002**, 124, 10274–10275.
- [12]  $[\alpha]_D^{15} = +39.7$  ( $c = 0.25$ ,  $\text{CHCl}_3$ ; literature:<sup>[1a]</sup>  $[\alpha]_D^{17} = +40.2$ ). The spectral data of the synthetic sample were identical with those of the natural product.
- [13] For the use of the modified Yamaguchi conditions, see: D. A. Evans, H. P. Ng, D. L. Rieger, *J. Am. Chem. Soc.* **1993**, 115, 11446–11459.
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